



**International Journal of Biology, Pharmacy
and Allied Sciences (IJBPAS)**
'A Bridge Between Laboratory and Reader'

www.ijbpas.com

**NEW ANTIMICROBIAL ACTIVE BIS-ISOFLAVONOID GLYCOSIDE FROM THE
ROOTS OF *CHLOROPHYTUM TUBER SUM ROXB.***

YADAV S* AND GUPTA S

Department of Chemistry, Bipin Bihari College, Jhansi, 284001, India

*Corresponding Author: E Mail: surabhiyadav1764@gmail.com;

Surabhiyadav_bbc@rediffmail.com

ABSTRACT

Phytochemical investigation of the roots of *Chlorophytum tuberosum Roxb* resulted in the isolation of a new Bis-isoflavonoid glycoside characterized as Bis(8-methyl-4'-methoxy-7-O- α -L-rhamnopyranoside)I-5,II-5 Bisisoflavone. Its structure has been determined by various chemical reactions as well as by using spectroscopic techniques like UV, IR, ^1H NMR, ^{13}C NMR, 2D NMR and FABMS. The new isolated bisisoflavanoid was found to possess significant antimicrobial and anti fungal activity.

Keywords: Antimicrobial Activity, *Chlorophytum tuberosum Roxb.*, Isoflavonoid Glycoside, Rhamnopyranoside

INTRODUCTION

The indigenous people of India use *Chlorophytum tuberosum* to cure various ailments along with its use in Aurveda, sidha and unani system of medicines [1]. Earlier workers have reported the isolation of polyphenolics, proteins, steroid from *Chlorophytum tuberosum* [2]. As a part of our investigation on the chemical constituents of medicinal plants here in we report the isolation and

structure elucidation of a new antimicrobial active isoflavone glycoside from the ethanolic extract of the roots of *Chlorophytum tuberosum Roxb.*

MATERIALS AND MEHODS

General

Melting point was determined in sulphuric acid-bath and uncorrected. the UV spectra were recorded on a unichem UV-2 spectrometer, the IR spectra were recorded on a perkin- elmer 842 IR

spectrophotometer in KBr pellets, the ^1H and ^{13}C NMR spectra were determined on brucker 300 spectrometer at 300 MHz respectively (chemical shift in δ ,ppm), instrument using DMSO- d_6 as solvent with TMS as an internal standard and HMBC experiments were carried out using conventional pulse sequence. The protonated FAB mass spectra were recorded on a JEOL SX 102/DA-6000 mass spectrometer/data system using Argon/Xenon (6KV, 10MA) as the FAB gas. Si gel GF $_{254}$ was used in TLC and spots were visualized by exposure over ammonia vapour and UV light as per requirement.

Plant Material

The roots of *Chlorophytum tuberosum* Roxb. were collected at Babai farm of Bundelkhand region of the district Jalaun. The specimen of the plants were deposited in herbarium of Bipin Bihari college and identified by staff of Botany department, Bipin Bihari College, Jhansi.

Extraction and Isolation

The roots of *Chlorophytum tuberosum* (Fam. Liliaceae) were dried and pulverized and (2.5kg) were extracted successively under reflux with ethanol at room temperature and dried to a thick

syrupey dark brown colour mass (80gm). The ethanolic extract was washed subsequently with benzene, ether and ethyl acetate. The ethyl acetate soluble part after concentration under vacuum chromatographed over silica gel (60-120 mesh). It was charged on sephadex LH-20 column (30 x 2.5 cm.). The column was eluted with ethyl acetate and methanol in different ratio. Obtained methanol extract (90:10) exhibited one spot by paper chromatography was subjected on silica gel (60-120 mesh) column chromatography. The gradient elution of column with chloroform and methanol solvent system in different ratios, results yielded compound-1 (75mg) from CHCl_3 :MeOH (30:70) and monitored by TLC.

Acid Hydrolysis of Compound-1

20mg compound in 10 ml ethanol and 35 ml of 7% alc. H_2SO_4 was taken in 100 ml round bottom flask fitted with water condenser & refluxed on a steam bath for 7 – 8 hours. After usual work up, the contents were concentrated and allowed to cool and the resultant mixture was extracted with Et_2O and concentrated to obtain aglycone part.

The aqueous hydrolysate after neutralization with BaCO_3 and

precipitate of BaSO₄ was filtered off. The filtrate was decolourised and lyophilized to concentrate. It was then examined over paper chromatography using n-BAW (4:1:5) as solvent system and aniline hydrogen phthalate as detecting agent, confirmed the presence of L-rhamnose (R_f=0.38) (Co-PC).

Antimicrobial Activity of Compound -1

The isolated bioflavonoid compound was screened for their antimicrobial activity by filter paper disc method [15], which were expressed in terms of diameter of inhibition zone.

The Potato Dextrose Agar method [15] were used to check antibacterial and antifungal activities respectively. Standard antibacterial and antifungal were also screened under similar conditions for comparison.

RESULTS AND DISCUSSION

The compound was isolated as a brown amorphous powder, m.p.218±2 °C, responded all the characteristic test of isoflavonoids and sugar, gave pink colour with Na-Hg/HCl, shinoda test [3, 4] indicating glycosidic isoflavonoid nature which was further confirmed by its UV and IR spectral data [5, 6]. Its molecular formula C₄₆H₄₆O₁₆, was deduced from molecular ion peak at [M+H]⁺ m/z 855, by protonated

FABMS. Its high molecular weight and the magnitude of the molar extinction coefficient value of compound 6.56, 6.6 led to conclude dimeric nature of biisoflavonoid [7].

Its IR spectrum showed a strong absorption bands at 2975 (C-H stretching), 2860(O-CH₃), 1625(C=O, non chelated), 1640(C=C, aromatic ring), 1392(C-Me bending), 1245(C-O-C stretching), 1183-1130 cm⁻¹(O-glycosidic linkage) [8].

The ¹H NMR spectrum (DMSO-d₆ at 300MHz) demonstrated the presence of characteristic sharp singlet in the higher downfield region of C-2 proton at δ_H 7.90 (2H,S) was indicative of two isoflavonoids units in compound [9]. The appearance of a singlet of two protons at δ_H 6.63(2H,s) indicated free protons at C-6 of A-ring of 1&2 unit. Further the availability of resonance for two ortho coupled protons forming two sets of AA' BB' spin system at δ_H 7.57(4H,s) and 7.28(4H,s) for H-2', 6', & H-3', 5' positions respectively. In the higher upfield region a singlet of 6 protons at δ_H 2.09(2x3H,s) was ascribable for two methyl groups in compound which were affixed at C-8 of A-ring with the availability of ²J

correlation [10]. A singlet at δ_H 3.90(2 x 3H,s) equivalent to 6 protons corroborated the presence of 2x OCH₃ groups at C-4' on B-ring of I&II unit [11]. The absence of chelated hydroxyl as well as hydrogen bonded proton in ¹H NMR spectrum of compound led to deduce interflavonyl linkage through C-5 (I,II) [12].

The anomeric proton signal at δ_H 5.33(2H, br, s) (I&II) was assigned to H-1'' of L-rhamnose, the remaining sugar protons resonated between δ_H 3.16-4.89, a triplet at δ_H 1.23(2x3H, t) in the upfield region of compound due to the two rhamnosyl methyl protons [13].

In ¹³C NMR spectrum, the chemical shift appeared in the upfield region at δ_C - 60.2 and 18.4 each of 6 protons intensity confirmed the presence of two methoxyl and methyl groups, assigning two positions at C-4' and C-8 respectively of both the unit, with complementary data of HMBC. The inter-isoflavonoid linkage between I&II isoflavonoid unit at C-5 was determined

by the presence of downfield shifting of quaternary carbon-carbon linkage at δ_C - 161.3 ppm(I&II) for C-5 [14]. the anomeric carbon signal at δ_C - 98.7 (I, II) was assigned to C-1'' of L-rhamnose sugar attached to C-7 of I&II unit of compound. Further in upfield region the presence of singlet at δ_C -17.5(I,II) was confirmatory of rhamnosyl methyl groups of sugar. Thus the C-7 site of the glycoside exhibited a shielding effect which confirmed that the attachment of sugar was at C-7 of the glycoside via a C-O-C linkage [14].

Complete ¹H and ¹³C NMR spectral assignments of compound was made with the aid of HMBC, HMQC experiments. The placements of CH₃ was established by the evidence of ²_J coupling correlation with quaternary carbon C-8 (δ_C - 104.1). On the basis of above discussion the structure of the compound was assigned as Bis (8-methyl-4'-methoxy-7-O- α -L-rhamnopyranoside) I-5, II-5 bisisoflavone (Figure 1).

Table 1: ¹H and ¹³C NMR Spectral Data of Compound -1

Position	Carbon(δ_H)	Proton (δ_H)	HMBC(C-H)
2 (I,II)	156.5	7.90(2H,S)	
3(I,II)	121.1	–	C-3,4
4(I,II)	177.8	–	
5 (I,II)	161.3	–	
6 (I,II)	98.7	6.63(2H,S)	
7 (I,II)	164.2	–	C-5,8
8 (I,II)	104.1	–	
9 (I,II)	157.3	–	
10 (I,II)	101.8	–	
1' (I,II)	121.1	–	
2' (I,II)	120.1	7.57 (2H,s)	
3' (I,II)	115.7	7.28 (2H,s)	C-3'
4' (I,II)	148.5	–	C-5'
5' (I,II)	115.7	7.28 (2H,s)	
6' (I,II)	120.1	7.57 (2H,s)	C-1',6'
CH ₃ at C-8 (I,II)	18.4	2.09 (2x3H,s)	C-1',4'
OCH ₃ at C-4' (I,II)	60.2	3.90 (2x3H,s)	C-8
			C-4'
Rhamnose	98.7	5.33 (2H,br,s)	
1'' (I,II)	70.6	4.89 (2H,d,J=6.0 Hz)	C-7
2'' (I,II)	70.4	4.61 (2H,s)	
3'' (I,II)	71.2	3.39 (2H,s)	
4'' (I,II)	70.1	3.16 (2H,s)	
5'' (I,II)			
(CH ₃) 6'' (I,II)	17.5	1.23 (2x3H, t)	

Table 2: Antibacterial Activity

S. NO.	Test Bacteria	Diameter of inhibition	Control*
1	<i>E. Coli</i> at 250 μ g/disc	12	18
2	<i>B.Subtilis</i> at 250 μ g/disc	14	24

*Streptomycin Against gm⁺ and gm⁻ Bacteria

Table 3: Antifungal Activity

S. NO.	Test Fungi	Diameter of inhibition	Control*
1	<i>Rhizopus stolonifer</i>	8	17
2	<i>Aspergillus niger</i>	7	15
3	<i>Penicillium expansum</i>	5	11

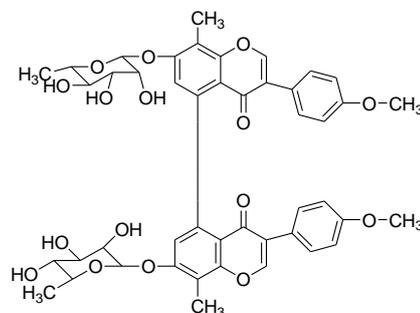
* β naphthol(20.0ppm)4'Methoxy 8 Methyl 7-O- α L Rhamnopyranoside bis Isoflavone

Figure 1

ACKNOWLEDGEMENT

The authors are thankful to Head, RSIC, CDRI, Lucknow for recording various spectra and also grateful to the Principal, Bipin Bihari College, Jhansi for providing laboratory facilities.

REFERENCES

- [1] Purohit SS, Aushadhiya and Sugandhit plant, Agrobios, Jodhpur India, 2004, 81-84.
- [2] Bhattacharjee S K and Michael AM, Hand Book of Medicinal Plants, Jaipur, India, 1998, 92-93.
- [3] Geissman TA, Modern method of plant analysis (ed) Peach K. Tracey M.V. Springer-Verleg Berlin, 1955.
- [4] Robinson T, The organic constituents of higher plants, Berges New York, 1962.
- [5] Mabry TJ, Markham KR and Thomas M, The Systematic Identification of Flavonoids, (springer, Berlin, Germany), 1970, 165-167.
- [6] Nakanishi K, Infra red Absorption Spectroscopy Practical, Holden day Inc. Sanfrancisco, 1962.
- [7] Jakson B, Locksley HD, Scheiman F and Wolstenholme WA, Extractives from Guttiferae, Part XXII. The Isolation and Structure of Four Novel Biflavanones from the Heartwoods of *Garcinia buchananii* Baker and *G. Eugeniifolia* Wall, J. Chem. Soc., (C) 1971, 3791-3804.
- [8] Li G, Zhao J, Tu Y, Yang X, Zhang H and Li L, Two new lignin glycosides from *Schisandra rubriflora*, Heterocycles, 63 (6), 2004, 1437-1441.
- [9] Singh OV, and Muthukrishnan M, Synthesis of isoflavones containing naturally occurring substitution pattern by oxidative rearrangement of respective flavanones using thallium (III) P-tosylate, Indian J. Chem., 44B, 2005, 2575-2581.
- [10] Tsompo A, Tene M, Kamnaing P, Ngnokam D, Ayafor JF, and Sterner O, Geranylated Flavonoids from *Dorstenia poinsettifolia*, Phytochem., 48 (2), 1998, 345-348.
- [11] Yadava RN, and Singh SK, New anti-inflammatory active flavanone glycoside from the *Echinops echinatus Roxby*, Indian J. Chem., 45B, 2006, 1004.
- [12] Yankep E, Mbafor JT, Fomum ZT, Steinbeck C, Messang BB, Nyasse B, Budzikiewicz H, Lenz C and Schmickler H, Phytochem., 56, 2001, 363-368.
- [13] Babu BR, Khurana S, Saluja R, Shrivastava AK, and Jain SS, A new flavone glycoside from

Zanthoxylum acanthopodium DC,
Indian J. of Chem., 46 (B), 2007,
872-874.

[14] Agrawal PK, NMR Spectroscopy in
Structural Elucidation of
oligosaccharides and glycosides,
Phytochem., 31, 1992, 3307-3330.

[15] Aneza KR, Experiments in
microbiology Plant pathology
Tissue culture and mushroom
cultivation, 2nd Ed., Visva
Prakashan – New Delhi, 1996.